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Area Under the Conc. Time Curve (AUC)

- The best measure for bioavailability (to measure the extent of absorption), could be:
 - > Model dependent: involves equation/model
 - ✓ We use integration to calculate AUC.

$$AUC = \frac{Xo}{K.Vd} = \frac{Xo}{Cl} = \frac{Co}{K}$$

Unit: Concentration.time (mg.hr/L)

- > Model-independent (<u>trapezoidal</u> rule): no equation / we use numerical integration
 - \checkmark Numerical integration: we divide the area into small sections and calculate their area.
 - ✓ When we divide the AUC (Area Under the Curve) into several sections, each section will resemble the shape of a trapezoid. The area of a trapezoid is calculated using the formula:

Area of trapezoid: $\frac{(Base1+Base2)}{2} * Height$

 \checkmark This formula helps in calculating the AUC by summing up the areas of these trapezoidal sections.



• Note:

 $\triangleright e^{-Kt}$: decay function.

 $\rightarrow -e^{-Kt}$: growth function.

Clearance (CL)

Definition: is the ratio of the rate of elimination by all routes to the concentration of the drug in plasma.

$$\succ CL = \frac{\text{Rate of elimination}\left(\frac{\text{mg}}{\text{h}}\right)}{\text{C in plasma}\left(\frac{\text{mg}}{\text{L}}\right)}$$

Unit: Volume/Time (L/h) or adjusted for body weight (l/h/kg)

$$Cl = K Vd$$

• Notes:

- There is no need for a negative sign (-) because it's a measurement of elimination (we already know it's a loss process).
- > The volume of serum or blood completely cleared of the drug per unit of time.
- It measures the elimination efficiency of all elimination organs (liver, kidney, gastrointestinal wall, lung, bile...) thus It's an additive parameter:

 $Cl_{total} = Cl_{RENAL} + Cl_{HEPATIC} + Cl_{PULMONARY} + \dots$

NOTE: usually we divide clearance into renal and non-renal

- Clearance (Cl) is the most important pharmacokinetic parameter because it determines the maintenance dose (MD) that is required to obtain a given steady-state serum concentration (Css): MD = Css
- Clearance itself is a model-independent parameter (we don't need to know the number of compartments to calculate clearance whether we calculated it through a dependent or independent way we will end with the same value).
- > So, the Clearance: the **volume of serum or blood** completely **cleared** from the drug **per unit time**.
- > The liver is most often the organ responsible for drug metabolism while in most cases the kidney is responsible for drug excretion.
- The gastrointestinal wall, lung, and kidney can also metabolize some drugs, and some medications are eliminated unchanged in the bile.
- Model dependent: Clearance = (dC/dt) /Cplasma <u>or</u> Clearance = K*Vd note: both rate and conc. in plasma are at the same time .

• Model-independent (trapezoidal): Using integration: Clearance = Xo / AUC

Cl total = Cl renal + Cl non-renalK*Vd = (Kr *Vd) + (Knr* Vd)= Vd (Kr + Knr)= Vd K

Kr: K for renal

Knr: K for non-renal

• Questions:

- > The units for clearance are:
 - A) Concentration/half-life
 - B) Dose/volume
 - C) Half-life/dose
 - D) Volume/time

Answer: D) Volume/time

> Total body clearance is the sum of clearance by the kidneys, liver, and other routes of elimination.

- A) True
- B) False

Answer: A) True.

To determine drug clearance, we must first determine whether a drug best fits a one- or twocompartment model.

- A) True
- C) False

Answer: B) False, because clearance Is a model-independent parameter

> Drug Y is given by an intravenous injection and plasma concentration are then determined as follows:

Time after	Conc (mg/L)
injection (hr)	
0	12
1	9.8
2	7.9
3	6.4
4	5.2
5	4.2
6	3.4
7	2.8
8	2.2





Is the drug eliminated by a first or zero order process? Defined your answer

Answer: We can plot time and conc. on both NORMAL and SEMI-LOG paper and examine:

If we get linearity on normal paper : zero order.

If we get linearity on semi-log paper: first order

But what if we don't have semi-log or normal paper? Simply, we know that the rate of reaction in first-order rxns changes as the concentration changes So the rate is decreasing (changing): first-order reaction

> Which of the following patient scenarios is associated with a smaller volume of distribution?

- A) Dose = 500 mg and initial serum concentration = 40 mg/L
- B) Dose = 20 mg and initial serum concentration = 1.5 mg/L

Answer: Vd=Xo/Co

- $A \rightarrow Vd=500/40=12.5 L$
- $B \rightarrow Vd=20/1.5=13.3 L$
- A is smaller volume of distribution.

- Explain how a person who weighs 70 kg can have a volume of distribution for a drug 700 L. Answer: Vd is apparent not physiological value so it can be larger than body volume (it describes the affinity of the drug)
- > For drug X, individual organ clearance have been determined as follows:

Renal clearance	180 mL/minute
Hepatic clearance	22 mL/minute
Pulmonary clearance	5.2 mL/minute

How would you describe the clearance of drug X?

Answer: Drug X is eliminated by three systems (renal, hepatic, and pulmonary), total CL=207.2mL/min but it's mainly eliminated by the renal system.

Cases in IV Bolus

• **Case 1:** Ten mg metoclopramide were administered intravenously to a 72 kg patient. The minimum plasma concentration required to cause significant enhancement of gastric emptying is 50ng/mL. The following plasma concentrations were observed after analysis of the specimen.

Time (hr)	Cp (ng/mL)
1	90
2	68
4	40
6	21.5
8	12
10	7





*Before starting the solution, let's summarize the information we have from the question: Dose (Xo) = 10 mg

Minimum effective concentration = 50 ng/ml \sim 50 mcg/L \sim 0.05 mg/L

> Plot the metoclopramide concentration-time data and draw a compartmental scheme showing the number of compartment involved.





- Write the equation describing the disposition kinetics of the drug. (Ct = Co e^{-Kt})
 - ✓ From the plot extrapolation Co = $120 \text{ ng/ml} \sim 120 \text{ mcg/L} \sim 0.120 \text{ mg/L}$.

✓ -K = slope =
$$\frac{\ln(90) - \ln(40)}{1 - 4}$$
 = -0.28 hr⁻¹

$$\checkmark$$
 Ct = 120 e -0.28t

\succ Calculate the biological half-life of the drug elimination (t¹/₂), the overall elimination rate-constant (K), the volume (Vd), the coefficient of distribution and the duration of action (td).

1)
$$t_{0.5} = \frac{0.693}{K} = \frac{0.693}{0.28} = 2.48 \text{ hr}$$

2) $K = 0.28 \text{ hr}^{-1}$
3) $Vd = \frac{Xo}{Co} = \frac{10 \text{ mg}}{0.12 \text{ mg/L}} = 83.3 \text{ L}$
4) Distribution coefficient $= \frac{83.3 \text{ L}}{72 \text{ kg}} = 1.157 \text{ L/Kg}$
5) Duration of action $= \text{Ct} = \text{Co e}^{-\text{Kt}}$
 $50 = 120 \text{ e}^{-0.28* \text{ td}}$
 $td = 3.13 \text{ hr}$
6) $AUC = \frac{Co}{K} = \frac{120 \text{ ng/mL}}{0.28 \text{ hr}^{-1}} = 425 \text{ ng.hr/Ml}$
7) $Cl = K \text{ Vd} = 0.28 \text{ hr}^{-1*} 83.3 \text{ L} = 23.5 \text{ L/hr}$

Comment on the extent of metoclopramide distribution in the body.

The patient weight 72, and the Vd= 83 (the volume of distribution is higher than the patient weight), it indicates that the drug has a strong affinity for tissues. The drug tends to distribute and accumulate in body tissues rather than staying primarily in the blood.

Case 2: An adult male patient was given the first dose of an antibiotic at 6:00 AM. At 12:00 noon the plasma level of the drug was measured and reported as 5 μ g/ml. The drug is known to follow the onecompartment model with a half-life of 6 hours. The recommended dosage regimen of this drug is 250 mg q.i.d. the minimum inhibitory concentration is 3 µg/ml. Calculate the following:

*Before starting the solution, let's summarize the information we have from the question:

$$6 \text{ am} \qquad 12 \text{ pm}$$

$$Xo = 250 \text{ mg} \qquad Xo = 250 \text{ mg} \qquad Ct = 5 \text{ mg/L}$$

$$t_{0.5} = 6 \text{ hr} \rightarrow \text{K} = 0.1155 \text{ hr}^{-1}$$

minimum effective conc = $3 \mu g/ml \sim 3 mg/L$

The apparent volume of distribution \geq

Solution 1:
$$Vd = \frac{Xo}{Co}$$

Xo = 250 mg

Since the half-life is 6 hours, and the concentration at 12 pm was 5 mg/L, the concentration 6 hours earlier would have been double that, 10 mg/L. This corresponds exactly to when we administered the first dose at 6 am.

$$Vd = \frac{250 mg}{10 mg/L} = 25 L.$$

Solution 2: find Co from the given data.

$$C = Co e^{-Kt}$$

$$5 = Co e^{-0.1155*6} \rightarrow Co = 10 \text{ mg/L}$$

$$Vd = \frac{250 \text{ mg}}{10 \text{ mg/L}} = 25 \text{ L}$$

> Expected plasma concentration at 10 AM.

At 10 AM -> 4 hours later $Ct = Co e^{-Kt}$ $Ct = 10 e^{-0.1155*4}$ Ct = 6.3 mg/L

> Duration of action of the first dose

 $C = 10 e^{-Kt}$ 3 = 10 $e^{-0.1155*td}$ td=10.4 hr

- Total body clearance Cl = K Vd Cl = 0.1155*25 Cl = 2.89 L/hr
- Fraction of the dose in the body 5 hours after the injection $X=Xo \ e^{-Kt}$ $\frac{X}{Xo} = e^{-0.1155*5} \rightarrow \frac{X}{Xo} = = 0.56$
- Total amount in the body 5 hours after the injection X=Xo e -Kt X=250 e -0.1155*5 X=140 mg
- Exponential and logarithmic equation (pharmacokinetic model) C = 10 e ^{-0.1155*t} LnC = 2.303 - 0.1155t LogC = 1 - 0.0502 t
- ➤ Total amount in the body immediately after injection of a second dose at 12:00 noon At 12 noon the first t_{0.5} has reached so → the first dose 250/2 = 125mg + we give a new dose 250mg X = 125mg + 250mg X = 375 mg
- > Duration of action of first dose only if dose administered at 6:00 AM was 500 mg.

Solution 1: $C = \frac{x_o}{v_d} = \frac{500 \text{ mg}}{25 \text{ L}} = 20 \text{ mg/L}$ $3=20^*e^{-0.1155^*td}$ td = 16.4 hr **Solution 2:** The difference between 500 and 250 represents one half-life. Therefore, doubling the dose in the case of first-order kinetics extends the duration of action by one half-life. The duration of action for 250 mg dose = 10.4 hr and the tage = 6 hr

the case of first-order kinetics extends the duration of action by one half-life. The duration of action for 250 mg dose = 10.4 hr and the $t_{0.5} = 6$ hr td = 10.4 + 6td = 16.4 hr Case 3: A general anesthetic has a volume of distribution of 15L and a minimum effective concentration of 2µg/mL (the drug is effective as long as the drug concentration is above 2µg/mL). After administration of 120mg of the drug as an IV bolus dose to a patient the drug produced an anesthetic effect for 6 h.

*Before starting the solution, let's summarize the information we have from the question: Vd = 15 L Co = Xo/Vd $C = Co e^{-Kt}$

 $\begin{array}{ll} \text{MEC} = 2 \text{ mg/L} & = 120/15 & 2 = 8 \text{ e}^{-6\text{K}} \\ \text{Xo} = 120 \text{ mg} & = 8 \text{ mg/L} & \text{K} = 0.23 \text{ hr} \\ \text{dt} = 6 \text{ hr} & \end{array}$

 $Ct = 8 e^{-0.23t}$

> Calculate the half-life of this drug.

Solution 1:
$$t_{0.5} = \frac{0.693}{K}$$

= $\frac{0.693}{0.23} \rightarrow t_{0.5} = 3 \text{ hr}$

<u>Solution 2:</u> The initial concentration (C₀) is 8. After one half-life ($t_{0.5}$), the concentration drops to 4, and after the second half-life, it drops to 2. After that, the drug no longer has an effect because the concentration falls below the MEC. Therefore, the drug remained effective for two half-lives.

The question mentions that the drug was active for 6 hours. Since it went through two half-lives in those 6 hours, we calculate: 6 hours / 2 half-lives = 3 hours per half-life.

So, the half-life is 3 hours.

- Calculate the minimum effective concentration for the drug if the dose was 400mg. The Minimum Effective Concentration is Constant for any drug and not affected by the dose.
- Calculate the expected duration of effect if an IV bolus dose of 240mg was administered. <u>Solution 1:</u>

 $X = Xo e^{-0.23t}$ 240 = 120 e^{-0.23*td} \rightarrow td = 9 hr

<u>Solution 2</u>: Doubling the dose will increase the duration of action by ONE $t_{0.5}$, The previous duration of action was six hours and the $t_{0.5}$ was 3 hours if we double the dose the duration of action will be: 6+3 = 9 hr

Calculate the lowest dose that will produce an effect for 3 h.

Solution 1: $Xt = 120 e^{-0.23t}$ $Xt = 120 e^{-0.23*3}$ Xt = 60 mg

Solution 2: 3 hrs is $t_{0.5}$ that we calculated before, the first dose was 120, after one $t_{0.5}$ it is : 60 mg

Calculate the expected duration of effect if 20mg was given as an IV bolus dose. If 20 mg were given, the conc. = 20mg/ 15L= 1.33 is below MEC !!! So duration of action is Zero. • **Case 4:** The therapeutic range of a drug is 20-200 mg/L. After an intravenous bolus injection of 1.0 gm followed by regression analysis of the concentration of the drug in plasma (in units of mg/L) versus time (in hours), the following linear equation was obtained:

Log Cp = 2 - 0.1t

*Before starting the solution, let's summarize the information we have from the question: MEC = 20 mg/L $Vd = \frac{x_0}{c_0}$ $\frac{K}{2.202} = 0.1$

	CU	2.303
Xo = 1000 mg	$Vd = \frac{1000}{100}$	K = 0.2303 hr ⁻¹
Log Co = 2	Vd = 10 L	$t_{0.5} = 3 hr$

Co = 100 mg

> Duration of action $20 = 100 e^{-0.2303td}$ t = 7 hr

t = 7 hr.

> Total body clearance

Cl = K Vd Cl = 0.2303*10Cl = 2.303 L/hr

Rate of elimination at 2 hours

To find the rate of elimination (dX/dt = KX), you first need to determine the amount of the drug remaining at 2-hour. $X = Xo e^{-Kt}$ $X = 1000 e^{-0.2303*2}$

X = 630.9 mg

$$\frac{dX}{dt} = KX = 0.2303*630.9 = 145.3 \text{ mg/hr}$$

Case 5: Drug X has a therapeutic range of 15-80 mg/L. After an intravenous bolus injection of 500 mg of drug X, the concentration of the drug in plasma (in units of μg/ml) versus time (in hours), were described by the following equation:

$Ct = 50 e^{-0.12t}$

*Before starting the solution, let's summarize the information we have from the question: MEC = 15 mg/L Xo = 500 mg Co = 50 mg/L Vd = 10 L $K = 0.12 \text{ hr}^{-1}$

Duration of action after the 500-mg dose. 15 = 50 e^{-0.12*td} t = 10 hr

Amount eliminated at 2 hours X = 500 e - 0.12 * 2X = 393.3 (remains) Eliminated = Total - Remain = 500 - 393.3Eliminated = 106.7 mgRate of elimination at 2 hours. \succ To find the rate of elimination (dX/dt = KX), you first need to determine the amount of the drug remaining at 2-hour. $X = Xo e^{-Kt}$ $X = 500 e^{-0.12*2}$ X = 393.3 mg $\frac{dX}{d} = KX$ = 0.12*393.3= 47.2 mg/hrCase 6: The plasma concentration-time profile after a single IV dose of 300mg of a drug was back extrapolated and the y-intercept was 7.5mg/L Xo = 300 mgCo = 7.5 mg/LCalculate the Vd of this drug. Vd = Xo/Co= 300/7.5Vd=40 LCalculate the dose that should achieve an initial drug concentration of 12mg/L. Xo = Co*Vd= 12*40Xo = 480 mgCase 7: After IV bolus administration of 450mg of a drug, the initial drug conc was found to be 15mg/L. Xo = 450 mgCo = 15 mg/LCalculate the Vd of this drug in this patient. $\mathbf{V}\mathbf{d} = \frac{Xo}{Co}$ 450 mg 15 mg/l Vd = 30 L

What is the IV bolus dose required to achieve an initial drug concentration of 20mg/L? Xo = Co*Vd = 20*30

- Xo = 600 mg
- > If the patient receives a single IV bolus dose of 2g of the drug, calculate the expected initial drug concentration after this large dose.

$$\mathbf{Co} = \frac{Xo}{Vd}$$
$$= \frac{2000 \ mg}{30 \ L}$$

Co = 66.7 mg/L



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